

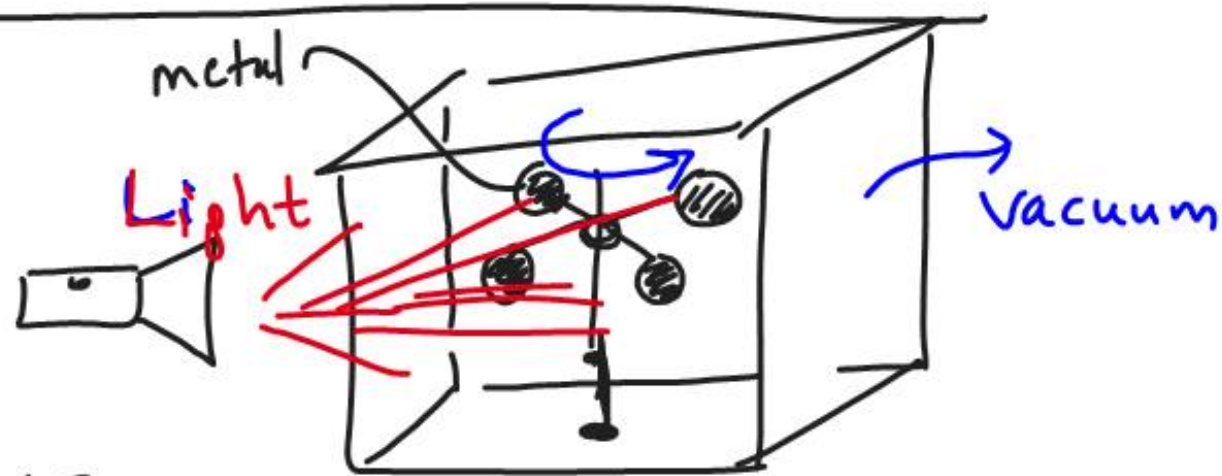
25 Sep 2012

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PHOTONICS & DEVICES (ECE 631)

PHOTON
is a particle
ZERO rest mass



$c = 3 \times 10^8$ m/s in vacuum

$$E = h \cdot \nu \quad \text{eV, J}$$

it has energy & momentum h : Planck's constant
 ν : frequency

Light intensity $\Rightarrow I$ (Irradiance)

$$I \Rightarrow (W/m^2)$$

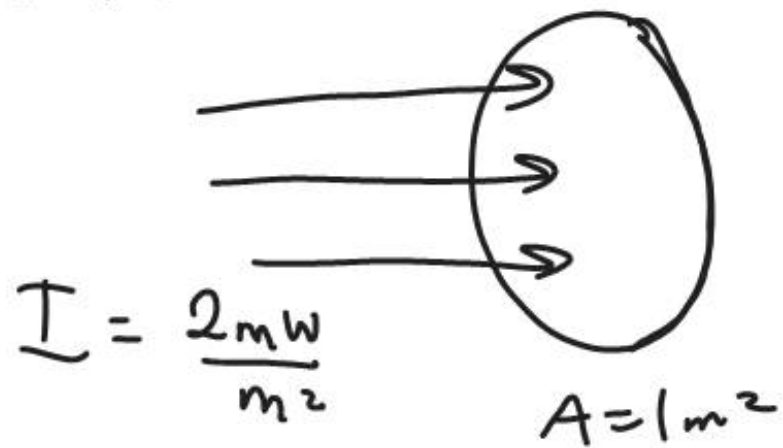
2 mW/m² intensity

$$\lambda = 1 \mu m$$

$$c = \lambda \cdot \nu$$

speed of light = $\nu \times \lambda$

how many photons we have in a $A = 1 m^2$
area



$$Power = 2 mW$$

$$Watt = \frac{\text{Joule/eV}}{\text{Sec}} = \frac{E}{\text{Sec}}$$

in 1 sec.

$$E = \underline{2 mJ}$$

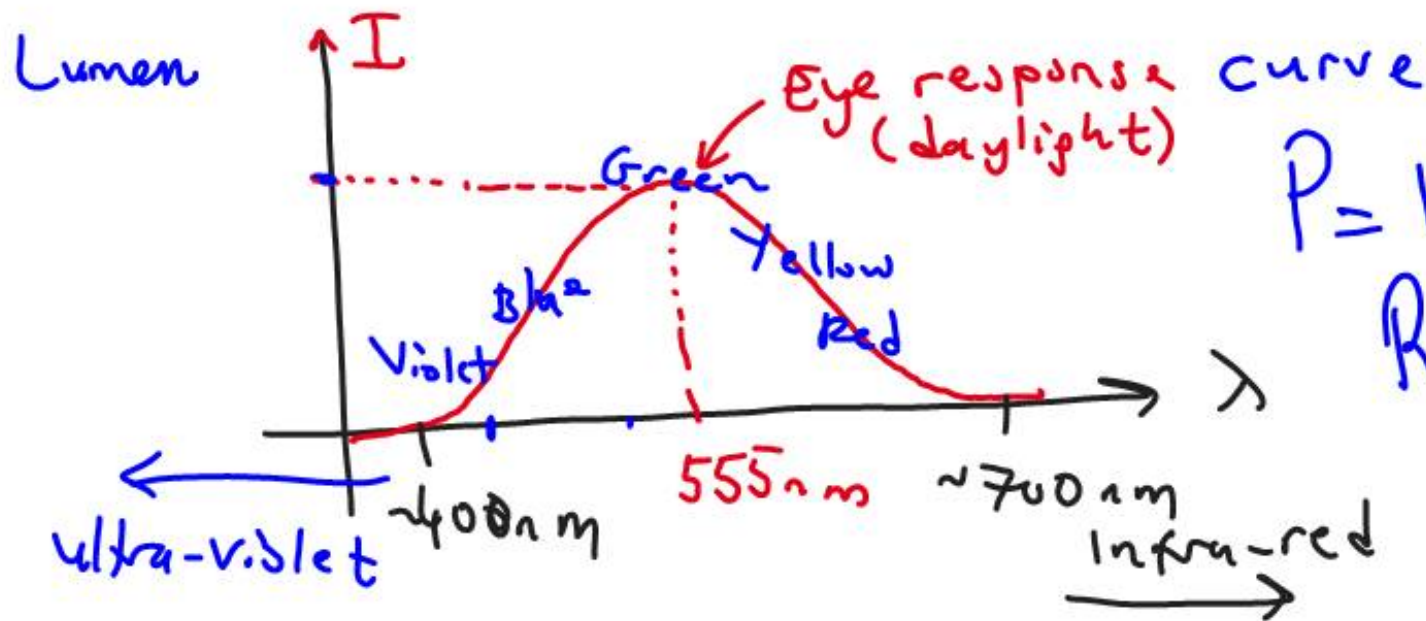
$$\text{Number of photons} = \frac{E}{h\nu}$$

$$\nu = \frac{3 \times 10^8 \text{ m/s}}{1 \times 10^{-6} \text{ m}} = 3 \times 10^{14} \text{ Hz}$$

$$E = h \cdot \nu = 6.626 \times 10^{-34} \text{ J}\cdot\text{s} \times 3 \times 10^{14} / \text{s}$$

$$E \approx 19.8 \times 10^{-20} \text{ J.}$$

$$\text{Number of photons} = \frac{2 \times 10^{-3} \text{ J}}{19.8 \times 10^{-20} \text{ J}} \approx \frac{10^{-2}}{10^{-20}} \approx \underline{10^{18} \text{ photons}}$$



$P = \text{Watt} \rightarrow \text{Lumen}$
 ↓
 Radiometry
 ↓
 photometry

Light \Rightarrow electromagnetic wave

THE ELECTRO MAGNETIC SPECTRUM

